

2014 Annual Drinking Water Quality Report

Bay County Wholesale Water System

We are pleased to report that our drinking water meets all federal and state requirements.

We are pleased to present to you this year's Annual Water Quality Report. This report is designed to inform you about the quality water and services we deliver to you every day. Our constant goal is to provide you with a safe and dependable supply of drinking water. We want you to understand the efforts we make to continually improve the water treatment process and protect our water resources. We are committed to ensuring the quality of your water. Our water source is surface water drawn from Deer Point Reservoir.

In 2014 the Department of Environmental Protection performed a Source Water Assessment on our system. The assessment was conducted to provide information about any potential sources of contamination in the vicinity of our surface water intakes. The surface water system is considered to be at high risk because of the many potential sources of contamination present in the assessment area. The assessment results are available on the FDEP Source Water Assessment and Protection Program website at www.dep.state.fl.us/swapp or they can be obtained from Bay County Utility Services by calling 850-248-5010.

The Bay County Water Treatment Plant uses a conventional treatment process consisting of coagulation, flocculation, sedimentation, filtration, pH adjustment, disinfection, fluoridation, and corrosion control. The treatment process includes adding lime occasionally to provide additional alkalinity to the raw water so it can react with the primary coagulating chemical, ferric sulfate, which is added to remove particles and organics. Polymer is also added to assist in the coagulation process. Sodium Hypochlorite is added to maintain disinfection in the distribution system. The addition of zinc orthophosphate reduces the corrosiveness of the water. Fluoride, in the form of hydrofluorosilicic acid, is added as a supplement to prevent tooth decay. Lime is also added at the end of the process to increase the pH. These processes are needed to meet the drinking water standards as set by the United States Environmental Protection Agency (EPA) and the Florida Department of Environmental Protection (FDEP).

We are pleased to report that our drinking water meets all federal and state requirements. If you have any questions about this report or concerning your water utility, please contact Donald Hamm, Water Division Superintendent at 850-248-5010. We encourage our valued customers to be informed about their water utility. If you want to learn more, please attend any of our regularly schedule Bay County Commission meetings. The meetings are scheduled the first and third Tuesday of each month. Public notices of the meetings are announced regularly publicizing the date, time, and location.

The Bay County Water Treatment Plant routinely monitors for contaminants in your drinking water according to Federal and State laws, rules, and regulations. Except where indicated otherwise, this report is based on the results of our monitoring for the period of January 1 to December 31, 2014. Data obtained before January 1, 2014, and presented in this report are from the most recent testing done in accordance with the laws, rules, and regulations.

In the table below, you may find unfamiliar terms and abbreviations. To help you better understand these terms we've provided the following definitions:

Maximum Contaminant Level or MCL: The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

Maximum Contaminant Level Goal or MCLG: The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Maximum residual disinfectant level or MRDL: The highest level of a disinfectant allowed in drinking water. There is convincing evidence that the addition of a disinfectant is necessary for control of microbial contaminants.

Maximum residual disinfectant level goal or MRDLG: The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

Nephelometric Turbidity Unit (NTU) - measure of the clarity of water. Turbidity in excess of 5 NTU is just noticeable to the average person.

"ND" means not detected and indicates that the substance was not found by laboratory analysis.

Parts per billion (ppb) or Micrograms per liter ($\mu g/l$) – one part by weight of analyte to 1 billion parts by weight of the water sample.

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Picocurie per liter (pCi/L) - measure of the radioactivity in water.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Microbiolog	jical Contamina	nts					
Contaminant and Unit of Measurement	Dates of sampling (mo./yr.)	MCL Violation Y/N	The Highest Single Measurement	The Lowest Monthly Percentage of Samples Meeting Regulatory Limits	MCLG	MCL	Likely Source of Contamination
Turbidity (NTU)	Jan 14 – Dec 14	N	0.40	99.4	N/A	TT	Soil runoff
Turbidity is a meas		the water. We	monitor it because it i	s a good indicator of	the effectivene	ss of our filtra	ion system. High turbidity
Turbidity is a meas	sure of the cloudiness of ctiveness of disinfectants	the water. We	monitor it because it i	s a good indicator of	the effectivene	ss of our filtra	ion system. High turbidity

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Contaminant and Unit of Measurement	Dates of sampling (mo. /yr.)	MCL Violation Y/N	Level Detected	Range of Results	MCLG	MCL	Likely Source of Contamination
Barium (ppm)	Apr-14	N	0.0086	N/A	2	2	Discharge of drilling wastes; discharge from metal refineries; erosion of natural deposits
Fluoride (ppm)	Apr-14	N	0.77	ND-0.77	4	4	Erosion of natural deposits; discharge from fertilizer and aluminum factories. Water additive which promotes strong teeth when at the optimum level of 0.7 ppm
Nickel (ppb)	Apr-14	N	2.4	N/A	N/A	100	Pollution from mining and refining operations. Natural occurrence in soil.
Nitrate (ppm)	Apr-14	N	0.204	N/A	10	10	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits
Sodium (ppm)	Apr-14	N	4.97	N/A	N/A	160	Salt water intrusion, leaching from soil

Stage 1 Disinfectants and Disinfection By-Products

Contaminant and Unit of Measurement	Dates of sampling (mo. /yr.)	MCL Violation Y/N	Level Detected	Range of Results	MCLG	MCL	Likely Source of Contamination
Chlorine (ppm)	Jan 14 – Dec 14	N	0.68	0.5 - 0.8	MRDLG = 4	MRDL = 4	Water additive used to control microbes
Contaminant and Unit of Measurement	Dates of sampling (mo/yr)	TT Violation Y/N	Lowest Running Annual Average, Computed Quarterly, of Monthly Removal Ratios	Range of Monthly Removal Ratios	MCLG	MCL	Likely Source of Contamination
Total Organic Carbon	Jan 14- Dec 14	N	1.74	1.59 – 2.24	N/A	П	Naturally present in the environment

Stage 2 Disinfectants and Disinfection By-Products

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Contaminant and Unit of Measurement	Dates of sampling (mo. /yr.)	MCL Violation Y/N	Level Detected	Range of Results	MCLG	MCL	Likely Source of Contamination
Haloacetic Acids (five) (HAA5) (ppb)	Jan 14 – Dec14	N	42.4	6.69-50.74	NA	MCL = 60	By-product of drinking water disinfection
TTHM [Total trihalomethanes] (ppb)	Jan 14 – Dec14	N	64.9	10.51-83.10	NA	MCL = 80	By-product of drinking water disinfection

Some people who drink water containing trihalomethanes in excess of the MCL, over many years, may experience problems with their liver, kidneys, or central nervous systems, and may have an increased risk of getting cancer.

We monitored for unregulated contaminants (UCs) in 2014 as part of a study to help the U.S. Environmental Protection Agency (EPA) determine the occurrence in drinking water of UCs and whether or not these contaminants need to be regulated. At present, no health standards (for example, maximum contaminant levels) or likely sources have been established for UCs. However, we are required to publish the analytical results of our UC monitoring in our annual water quality report. If you would like more information on the EPA's Unregulated Contaminants Monitoring Rule, please call the Safe Drinking Water Hotline at (800) 426-4791.

Unregulated Contaminants

Contaminant and Unit of Measurement	Dates of sampling (mo. /yr.)	Detected Levels	Range of Results	Likely Source of Contamination
Strontium (ppb)	Jan 14 – July 14	58.25	50.0 - 66.5	Unavailable
Chlorate (ppb)	Jan 14 – July 14	173.0	93.1 - 315	Unavailable
Chromium-6 (ppb)	Jan 14 – July 14	0.019	ND - 0.038	Unavailable
Vanadium (ppb)	Jan 14 – July 14	0.07	ND - 0.21	Unavailable

Lead and Copper (Tap Water)

Contaminant and Unit of Measurement	Dates of sampling (mo./yr.)	AL Violation Y/N	90th Percentile Result	No. of sampling sites exceeding the AL	MCLG	AL (Action Level)	Likely Source of Contamination
Copper (tap water) (ppm)	Aug 14	N	0.404	0 of 30	1.3	1.3	Corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Lead (tap water) (ppb)	Aug 14	N	0.7	0 of 30	0	15	Corrosion of household plumbing systems; erosion of natural deposits

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Bay County Utility Services is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

- (A) Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- (B) Inorganic contaminants, such as salts and metals, which can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- (C) Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- (D) Organic chemical contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, and septic systems.
- (E) Radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the EPA prescribes regulations, which limit the amount of certain contaminants in water provided by public water systems. The Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water, which must provide the same protection for public health.

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the Environmental Protection Agency's Safe Drinking Water Hotline at 1-800-426-4791.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. EPA/CDC guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbiological contaminants are available from the Safe Drinking Water Hotline (800-426-4791).

We at the Bay County Water Treatment Plant work around the clock to provide top quality water to every tap. We ask that all our customers help us protect our water sources, which are the heart of our community, our way of life and our children's future.

2014 Annual Drinking Water Quality Report Air Force Research Laboratory, Tyndall AFB

This year's Annual Water Quality Report covering CY 2014 is provided to you by Gulf Coast Electric Cooperative. The water distribution system was privatized under Contract SP0600-10-C-8253 on June 1, 2011, thus all sampling before this date was performed by Bioenvironmental Engineering Flight. Subsequent testing after June 1, 2011 was performed by Gulf Coast Electric Cooperative. This report is designed to inform you about the water quality and services that are delivered to you every day. This report was prepared and distributed in accordance with Air Force Instruction 48-144, Safe Drinking Water Surveillance Program. We are committed to ensuring the quality of your water.

The Air Force Research Laboratory (AFRL) water system is classified as non-transient non-community, which means that it is a public water system that regularly serves at least 25 of the same persons over six months per year. The drinking water source for the AFRL is a well which draws from the Floridan Aquifer, a groundwater source. The well is located in Building 9705; access to this well is highly restricted, greatly reducing the risk of intentional contamination. Immediately upon being pumped from the aquifer, water is disinfected with chlorine to protect against microbial contamination.

On June 13, 2012 the Florida Department of Environmental Protection (FDEP) conducted an inspection of the AFRL public water system. The purpose of this survey was to determine the system's capability to provide an adequate potable water supply that complies with the Florida Safe Drinking Water Act. The system was found to be in good operational order. Recommendations were to test the flow meter for accuracy. The test flow meter was checked by Florida Rural Water Association and was found to be accurately working.

Gulf Coast Electric Cooperative routinely monitors for contaminants in your drinking water according to Federal and State laws, rules, and regulations. Except where indicated otherwise, this report is based on the results of our monitoring for the period of January 1 to December 31, 2014. Data obtained before January 1, 2012, and presented in this report are from the most recent testing done in accordance with the laws, rules, and regulations.

This report shows our water quality results and what they mean. If you have any questions about this report or concerning your water utility, please contact Gulf Coast Electric Cooperative at 850-481-1188. We encourage our valued customers to be informed about their water utility.

The following tables list monitoring frequency for contaminants and applicable test results for the 2014 calendar year. We constantly monitor for various contaminants in the water supply to meet all regulatory requirements. The tables list only those contaminants that had some level of detection. Many other contaminants have been analyzed, but were not present or were below the detection limits of the lab equipment. Detection does not necessarily mean that the contaminant exceeded its Maximum Contaminant Level (MCL) or that its presence in the drinking water poses a health risk. The state allows us to monitor for certain contaminants less than once per year because the concentration of these contaminants do not change frequently. The inclusion of this data is required in water quality reports until more current sampling is requested and analyzed. As shown by the dates of sampling, some of our data, though

representative of the water quality, is more than one year old. In the tables below, you may find unfamiliar terms and abbreviations. To help you better understand these terms we've provided the following definitions:

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Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Initial Distribution System Evaluation (IDSE): An important part of the Stage 2 Disinfection Byproducts Rule (DBPR). The IDSE is a one-time study conducted by water systems to identify distribution system locations with high concentrations of trihalomethanes (THMs) and haloacetic acids (HAAs). Water systems will use results from the IDSE, in conjunction with their Stage 1 DBPR compliance monitoring data, to select compliance monitoring locations for the Stage 2 DBPR.

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Picocurie per liter (pCi/L) - measure of the radioactivity in water.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Analyte Groups and Monitoring Frequency Table

Analyte/Contaminant Group	Monitoring Frequency
Biological Contaminants	Monthly
Nitrate & Nitrite	Annually
Inorganic	Every 3 years (Analyzed in 2013)
Volatile Organics	Every 3 years (Analyzed in 2013)
Synthetic Organic Contaminants	Every 3 years (Analyzed in 2013)
Trihalomethanes and Haloacetic Acids	Every 3 years (Analyzed in 2014)
Lead and Copper	Every 3 years - 5 samples (Analyzed in 2012)
Radiochemical	No longer required

2014 Contaminant Tables

Test Results for Inorganic Compounds								
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MCL	Likely Source of Contamination		
Barium (mg/l)	8/20/13	N	0.015	2	2	Erosion of natural deposits		
Sodium (mg/l) (Inorganic Analyte)	8/20/13	N	46	N/A	160	Salt water intrusion, leaching from soil		
Fluoride (mg/l)	8/20/13	Y	5.0*	4	4	Erosion of natural deposits		
Nitrate (mg/l)	9/15/14	N	0.1	10	10	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits		
Nitrite (mg/l)	9/15/14	N	0.1	1	1	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits		

^{*} Due to historical Fluoride results, the FDEP requires the use of Reverse Osmosis units for potability use. Please see notes on Fluoride Sampling discussion below.

Test Results for Disinfectant/Disinfection By-Product (D/DBP) Parameters (2)								
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MCL	Likely Source of Contamination		
Chlorine (ppm)	Jan – Dec 14	N	0.2 - 3.0	4	4	Water additive used to control microbes		
Haloacetic Acids (five) (HAA5) (ug/l)	9/15/14	N	4.19	N/A	60	By-product of drinking water disinfection		
TTHM (Total Trihalomethanes) (ug/l)	9/15/14	N	11.6	N/A	80	By-product of drinking water disinfection		

Test Results for Radiochemical Contaminants								
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MC	L	Likely Source of Contamination	
Gross Alpha (pCi/L)	6/19/98	N	2.3	N/A	15		Erosion of natural deposits	
	Test Results for Lead and Copper							
Contaminant and Unit of Measurement	Sampling Date	AL Violation Y/N	90th Percentile Result	# of Samples Exceeding MCL	MCLG	AL	Likely Source of Contamination	
Lead (mg/l)	8/17/12	N	0.1	0 of 5	0.015	0.015	Corrosion of household plumbing systems; erosion of natural deposits	
Copper (mg/l)	8/17/12	N	0.07	0 of 5	1.36	1.36	Corrosion of household plumbing systems, erosion of natural denosits	

Bacteriological Sampling at Air Force Research Laboratory

From January 1, 2014 to December 31, 2014, Gulf Coast Electric Cooperative conducted monthly microbiological sampling at a variety of points at AFRL. During 2014, zero samples tested positive for bacteriological contamination.

Fluoride Sampling at Air Force Research Laboratory

Fluoride is a FDEP secondary drinking standard whose level is set at 4.0 mg/l. High levels of fluoride are caused by erosion of natural deposits in ground water. Past Inorganic Compounds testing results show high levels of fluoride in the AFRL area. To minimize these levels, AFRL is required to use Reverse Osmosis Units for any potable water requirements. The Bioenvironmental Engineering Flight performs periodic fluoride sampling on these units to ensure that they are operating adequately.

Lead Sampling at AFRL

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Gulf Coast Electric Cooperative is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

Health Effects of Detected Contaminants

Barium - Some people who drink water containing barium in excess of the MCL over many years could experience an increase in their blood pressure.

Fluoride - Some people who drink water containing fluoride in excess of the MCL over many years could get bone disease, including pain and tenderness of the bones. Fluoride in drinking water at half the MCL or more may cause mottling of children's teeth, usually in children less than nine years old. Mottling also known as dental fluorosis, may include brown staining and/or pitting of the teeth, and occurs only in developing teeth before they erupt from the gums.

Sodium - The FDEP has set the drinking water standard for sodium at 160.0 parts per million (ppm) to protect individuals that are susceptible to sodium sensitive hypertension or diseases that cause difficulty in regulating body fluid volume. Sodium is monitored so that individuals who have been placed on sodium (salt) restricted diets may take the sodium in their water into account. Sodium naturally occurs in food and drinking water. Food is the common source of sodium. Drinking water contributes only a small fraction (less than 10 percent) to the overall sodium intake. Sodium levels in drinking water can be increased by ion-exchange softeners at water treatment facilities or some point-of-use treatment devices.

Haloacetic Acids (HAA) - Some people who drink water containing haloacetic acids in excess of the MCL over many years may have an increased risk of getting cancer.

Nitrate - Infants below the age of six months who drink water containing nitrate in excess of the MCL could become seriously ill and, if untreated, may die. Symptoms include shortness of breath and blue baby syndrome.

Nitrite - Infants below the age of six months who drink water containing nitrite in excess of the MCL could become seriously ill and, if untreated, may die. Symptoms include shortness of breath and blue baby syndrome.

TTHMs (Total Trihalomethanes) - Some people who drink water containing trihalomethanes in excess of the MCL over many years may experience problems with their liver, kidneys, or central nervous systems, and may have an increased risk of getting cancer.

Alpha emitters - Certain minerals are radioactive and may emit a form of radiation known as alpha radiation. Some people who drink water containing alpha emitters in excess of the MCL over many years may have an increased risk of getting cancer.

Radium - Some people who drink water containing radium 226 or 228 in excess of the MCL over many years may have an increased risk of getting cancer.

Lead - Infants and children who drink water containing lead in excess of the action level could experience delays in their physical or mental development. Children could show slight deficits in attention span and learning abilities. Adults who drink this water over many years could develop kidney problems or high blood pressure.

Copper - Copper is an essential nutrient, but some people who drink water containing copper in excess of the action level over a relatively short amount of time could experience gastrointestinal distress. Some people who drink water containing copper in excess of the action level for many years could suffer liver or kidney damage. People with Wilson's disease should consult their personal doctor.

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Nitrate & Nitrite	Annually
Inorganic	Every 3 years (Analyzed in 2013)
Volatile Organics	Every 3 years (Analyzed in 2013)
Synthetic Organic Contaminants	Every 3 years (Analyzed in 2013)
Trihalomethanes and Haloacetic Acids	Every 3 years (Analyzed in 2014)
Lead and Copper	Every 3 years - 5 samples (Analyzed in 2012)
Radiochemical	No longer required

2014 Contaminant Tables

Test Results for Inorganic Compounds									
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MCL	Likely Source of Contamination			
Barium (mg/l)	8/20/13	N	0.039	2	2	Erosion of natural deposits			
Fluoride (mg/l) (Inorganic Analyte)	8/20/13	N	2.6	4	4	Erosion of natural deposits			
Sodium (mg/l) (Inorganic Analyte)	8/20/13	N	110	N/A	160	Salt water intrusion, leaching from soil			
Nitrate (mg/l)	9/15/14	N	0.01	10	10	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits			
Nitrite (mg/l)	9/15/14	N	0.01	1	1	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits			

Test Results for Disinfectant/Disinfection By-Product (D/DBP) Parameters (2)										
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MCL	Likely Source of Contamination				
Chlorine (ppm)	Jan – Dec 14	N	0.3 - 2.5	4	4	Water additive used to control microbes				
Haloacetic Acids (five) (HAA5) (ug/l)	9/15/14	N	4.2	NA	60	By-product of drinking water disinfection				
TTHM (Total trihalomethanes) (ug/l)	9/15/14	N	0.58	NA	80	By-product of drinking water disinfection				

Test Results for Radiochemical Contaminants									
Contaminant and Unit of Measurement Sampling Date Violation Y/N Level Detected MCLG MCL Likely Source of Contamination									
Gross Alpha (pCi/L)	4/7/98	N	5.3	N/A	15	Erosion of natural deposits			

Test Results for Lead and Copper										
Contaminant and Unit of Measurement	Sampling Date	AL Violation Y/N	90th Percentile Result	# of Samples Exceeding MCL	MCLG	AL	Likely Source of Contamination			
Lead (mg/l)	8/17/12	N	0.001	0 of 5	0.015	0.015	Corrosion of household plumbing systems; erosion of natural deposits			
Copper (mg/l)	8/17/12	N	0.21	0 of 5	1.36	1.36	Corrosion of household plumbing systems, erosion of natural deposits			

Bacteriological Sampling at Full Scale Drone

From January 1, 2014 to December 31, 2014, Gulf Coast Electric Cooperative conducted monthly microbiological sampling at a variety of points at Full Scale. During 2014, zero samples tested positive for bacteriological contamination.

Health Effects of Detected Contaminants

Barium - Some people who drink water containing barium in excess of the MCL over many years could experience an increase in their blood pressure.

Fluoride - Some people who drink water containing fluoride in excess of the MCL over many years could get bone disease, including pain and tenderness of the bones. Fluoride in drinking water at half the MCL or more may cause mottling of children's teeth, usually in children less than nine years old. Mottling also known as dental fluorosis, may include brown staining and/or pitting of the teeth, and occurs only in developing teeth before they erupt from the gums.

Sodium - The FDEP has set the drinking water standard for sodium at 160.0 parts per million (ppm) to protect individuals that are susceptible to sodium sensitive hypertension or diseases that cause difficulty in regulating body fluid volume. Sodium is monitored so that individuals who have been placed on sodium (salt) restricted diets may take the sodium in their water into account. Sodium naturally occurs in food and drinking water. Food is the common source of sodium. Drinking water contributes only a small fraction (less than 10 percent) to the overall sodium intake. Sodium levels in drinking water can be increased by ion-exchange softeners at water treatment facilities or some point-of-use treatment devices.

Haloacetic Acids (HAA) - Some people who drink water containing haloacetic acids in excess of the MCL over many years may have an increased risk of getting cancer.

Nitrate - Infants below the age of six months who drink water containing nitrate in excess of the MCL could become seriously ill and, if untreated, may die. Symptoms include shortness of breath and blue baby syndrome.

Nitrite - Infants below the age of six months who drink water containing nitrite in excess of the MCL could become seriously ill and, if untreated, may die. Symptoms include shortness of breath and blue baby syndrome.

TTHMs (Total Trihalomethanes) - Some people who drink water containing trihalomethanes in excess of the MCL over many years may experience problems with their liver, kidneys, or central nervous systems, and may have an increased risk of getting cancer.

Alpha emitters - Certain minerals are radioactive and may emit a form of radiation known as alpha radiation. Some people who drink water containing alpha emitters in excess of the MCL over many years may have an increased risk of getting cancer.

Radium - Some people who drink water containing radium 226 or 228 in excess of the MCL over many years may have an increased risk of getting cancer.

Lead - Infants and children who drink water containing lead in excess of the action level could experience delays in their physical or mental development. Children could show slight deficits in attention span and learning abilities. Adults who drink this water over many years could develop kidney problems or high blood pressure.

Copper - Copper is an essential nutrient, but some people who drink water containing copper in excess of the action level over a relatively short amount of time could experience gastrointestinal distress. Some people who drink water containing copper in excess of the action level for many years could suffer liver or kidney damage. People with Wilson's disease should consult their personal doctor.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

Contaminants that may be present in source water include:

a. Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.

- b. Inorganic contaminants, such as salts and metals, which can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- c. Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- d. Organic chemical contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, and septic systems.
- e. Radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the EPA prescribes regulations, which limit the amount of certain contaminants in water provided by public water systems. The Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water, which must provide the same protection for public health.

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the Bioenvironmental Engineering Flight at 850-283-7139 or the Environmental Protection Agency's Safe Drinking Water Hotline at 1-800-426-4791.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. EPA/CDC guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbiological contaminants are available from the Safe Drinking Water Hotline (800-426-4791).

2014 Annual Drinking Water Quality Report Silver Flag, Tyndall AFB

This year's Annual Water Quality Report covering CY 2014 is provided to you by Gulf Coast Electric Cooperative and Bioenvironmental Engineering Flight. The water distribution system was privatized under Contract SP0600-10-C-8253 on June 1, 2011, thus all sampling before this date was performed by Bioenvironmental Engineering Flight. Subsequent testing after June 1, 2011 was performed by Gulf Coast Electric Cooperative. This report is designed to inform you about the water quality and services that are delivered to you every day. This report was prepared and distributed in accordance with Air Force Instruction 48-144, Safe Drinking Water Surveillance Program. We are committed to ensuring the quality of your water

The Silver Flag water system is classified as non-transient non-community, which means that it is a public water system that regularly serves at least 25 of the same persons over six months per year. The drinking water source for Silver Flag is a well which draws from the Floridan Aquifer, a groundwater source. The well is located in Building 9439 in the Red Horse Training Compound; access to this well is highly restricted, greatly reducing the risk of intentional contamination. Immediately upon being pumped from the aquifer, water is disinfected with chlorine to protect against microbial contamination.

On June 13, 2012 the Florida Department of Environmental Protection (FDEP) conducted an inspection of the Silver Flag public water system. The purpose of this survey was to determine the system's capability to provide an adequate potable water supply that complies with the Florida Safe Drinking Water Act. The system was found to be in good operational order. Recommendations were to test the flow meter for accuracy. The test flow meter was checked by Florida Rural Water Association and was found to be accurately working.

Gulf Coast Electric Cooperative routinely monitors for contaminants in your drinking water according to Federal and State laws, rules, and regulations. Except where indicated otherwise, this report is based on the results of our monitoring for the period of January 1 to July 11, 2014. Data obtained before January 1, 2012, and presented in this report are from the most recent testing done in accordance with the laws, rules, and regulations.

This report shows our water quality results and what they mean. If you have any questions about this report or concerning your water utility, please Gulf Coast Electric Cooperative at 850-481-1188. We encourage our valued customers to be informed about their water utility.

The following tables list monitoring frequency for contaminants and applicable test results for the 2014 calendar year. We constantly monitor for various contaminants in the water supply to meet all regulatory requirements. The tables list only those contaminants that had some level of detection. Many other contaminants have been analyzed, but were not present or were below the detection limits of the lab equipment. Detection does not necessarily mean that the contaminant exceeded its Maximum Contaminant Level (MCL) or that its presence in the drinking water poses a health risk. The state allows us to monitor for certain contaminants less than once per year because the concentration of these contaminants do not change frequently. The inclusion of this data is required in water quality reports until more current sampling is requested and analyzed. As shown by the dates of sampling, some of our data, though representative of the water quality, is more than one year old.

In the tables below, you may find unfamiliar terms and abbreviations. To help you better understand these terms we've provided the following definitions:

Maximum Contaminant Level or MCL: The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

Maximum Contaminant Level Goal or MCLG: The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

Initial Distribution System Evaluation (IDSE): An important part of the Stage 2 Disinfection Byproducts Rule (DBPR). The IDSE is a one-time study conducted by water systems to identify distribution system locations with high concentrations of trihalomethanes (THMs) and haloacetic acids (HAAs). Water systems will use results from the IDSE, in conjunction with their Stage 1 DBPR compliance monitoring data, to select compliance monitoring locations for the Stage 2 DBPR.

Maximum residual disinfectant level or MRDL: The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum residual disinfectant level goal or MRDLG: The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

Nephelometric Turbidity Unit (NTU): A measure of the clarity of water. Turbidity in excess of five NTU is just noticeable to the average person.

"ND" means not detected and indicates that the substance was not found by laboratory analysis.

Parts per billion (ppb) or Micrograms per liter (μ g/I) – one part by weight of analyte to 1 billion parts by weight of the water sample.

Parts per million (ppm) or Milligrams per liter (mg/l) – one part by weight of analyte to 1 million parts by weight of the water sample.

Picocurie per liter (pCi/L) - measure of the radioactivity in water.

Treatment Technique (TT): A required process intended to reduce the level of a contaminant in drinking water.

Analyte Groups and Monitoring Frequency Table

Analyte/Contaminant Group	Monitoring Frequency
Biological Contaminants	Monthly
Nitrate & Nitrite	Annually
Inorganic	Every 3 years (Analyzed in 2013)
Volatile Organics	Every 3 years (Analyzed in 2013)
Synthetic Organic Contaminants	Every 3 years (Analyzed in 2013)
Trihalomethanes and Haloacetic Acids	Every 3 years (Analyzed in 2013)
Lead and Copper	Every 3 years - 5 samples (Analyzed in 2012)
Radiochemical	No longer required

2014 Contaminant Tables

Test Results for Inorganic Compounds										
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MCL	Likely Source of Contamination				
Barium (mg/l)	8/20/13	N	0.02	2	2	Erosion of natural deposits				
Sodium (mg/l) (Inorganic Analyte)	8/20/13	N	72	N/A	160	Salt water intrusion, leaching from soil				
Fluoride (mg/l)	8/20/13	N	3.2	4	4	Erosion of natural deposits				
Nitrate (mg/l)	8/20/13	N	0.01	10	10	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits				
Nitrite (mg/l)	8/20/13	N	0.01	1	1	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits				

^{*} Due to historical Fluoride results, the FDEP requires the use of Reverse Osmosis units for potability use. Please see notes on Fluoride Sampling discussion below.

Test Results for Disinfectant/Disinfection By-Product (D/DBP) Parameters (2)										
Contaminant and Unit of Measurement	Sampling Date	MCL Violation Y/N	Level Detected	MCLG	MCL	Likely Source of Contamination				
Chlorine (ppm)	Jan – July 14	N	0.2 - 3.0	4	4	Water additive used to control microbes				
Haloacetic Acids (five) (HAA5) (ug/l)	8/22/13	N	2.31	NA	60	By-product of drinking water disinfection				
TTHM (Total trihalomethanes) (ug/l)	8/22/13	N	14.2	NA	80	By-product of drinking water disinfection				

Test Results for Radiochemical Contaminants									
Contaminant and Unit of Measurement Sampling Date WCL Violation Y/N Level Detected MCL Likely Source of Contamination									
Gross Alpha (pCi/L)	6/16/98	N	4.1	N/A	15	Erosion of natural deposits			
Radium 226 (pCi/L)	6/16/98	N	1.2	N/A	20	Erosion of natural deposits			
Radium 228 (pCi/L)	6/16/98	N	0.1	N/A	20	Erosion of natural deposits			

Test Results for Lead and Copper										
Contaminant and Unit of Measurement	Sampling Date	AL Violation Y/N	90th Percentile Result	# of Samples Exceeding MCL	MCLG	AL	Likely Source of Contamination			
Lead (mg/l)	8/14/12	N	0.0018	0 of 5	0.015	0.015	Corrosion of household plumbing systems; erosion of natural deposits			
Copper (mg/l)	8/14/12	N	0.236	0 of 5	1.36	1.36	Corrosion of household plumbing systems, erosion of natural deposits			

Bacteriological Sampling at Silver Flag

From January 1, 2014 to July 11, 2014, Gulf Coast Electric Cooperative conducted monthly microbiological sampling at a variety of points at Silver Flag. During April total coliforms were detected but were resolved during second sampling.

Fluoride Sampling at Silver Flag

Fluoride is a FDEP secondary drinking standard whose level is set at 4.0 mg/l. High levels of fluoride are caused by erosion of natural deposits in ground water. Past Inorganic Compounds testing results show high levels of fluoride in the Silver Flag area. To minimize these levels, Silver Flag is required to use Reverse Osmosis Units for any potable water requirements. The Bioenvironmental Engineering Flight performs periodic fluoride sampling on these units to ensure that they are operating adequately.

Lead Sampling at Silver Flag

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Tyndall AFB Utility Services is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at http://www.epa.gov/safewater/lead.

Health Effects of Detected Contaminants

Barium - Some people who drink water containing barium in excess of the MCL over many years could experience an increase in their blood pressure.

Fluoride - Some people who drink water containing fluoride in excess of the MCL over many years could get bone disease, including pain and tenderness of the bones. Fluoride in drinking water at half the MCL or more may cause mottling of children's teeth, usually in children less than nine years old. Mottling also known as dental fluorosis, may include brown staining and/or pitting of the teeth, and occurs only in developing teeth before they erupt from the gums.

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the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

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